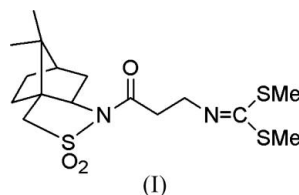
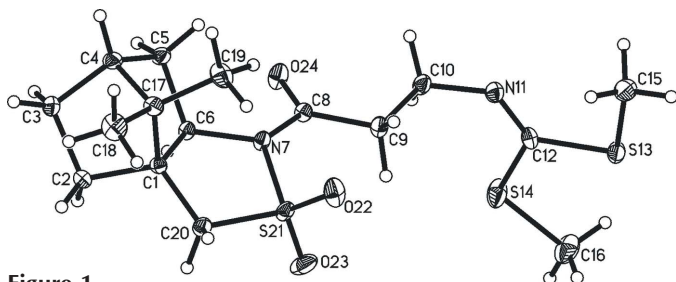


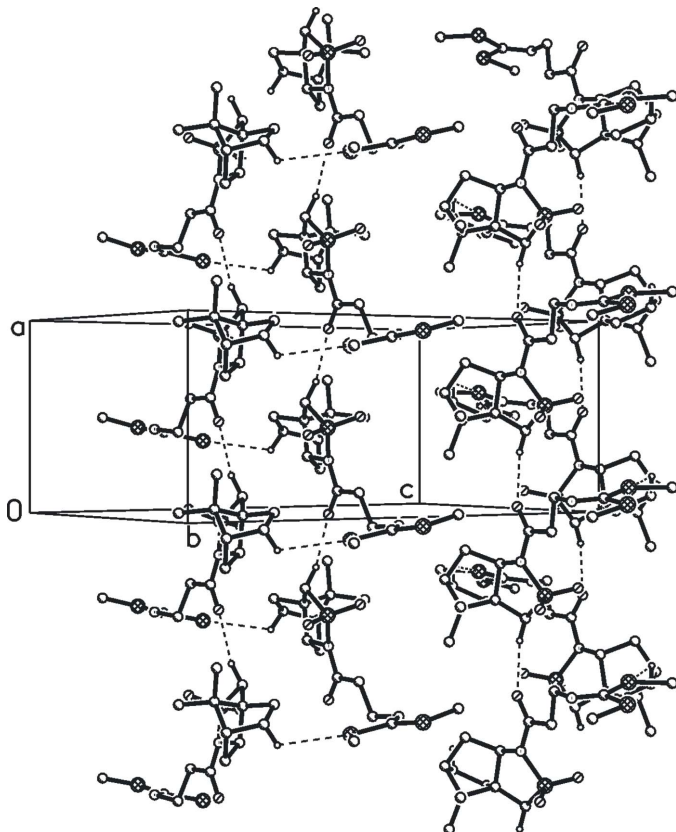
Dagmara Woźniak,<sup>a</sup> Aneta  
Szymańska,<sup>a</sup> Artur Sikorski,<sup>a\*</sup>  
Antoni Konitz<sup>a,b</sup> and  
Tadeusz Lis<sup>c</sup><sup>a</sup>University of Gdańsk, Faculty of Chemistry,  
J. Sobieskiego 18, 80-952 Gdańsk, Poland,  
<sup>b</sup>Gdańsk University of Technology, Department  
of Inorganic Chemistry, G. Narutowicza 11/12,  
80-952 Gdańsk, Poland, and <sup>c</sup>University of  
Wrocław, Faculty of Chemistry, F. Joliot-Curie  
14, 50-383 Wrocław, Poland

Correspondence e-mail: art@chem.univ.gda.pl

**Key indicators**Single-crystal X-ray study  
 $T = 100$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.025  
 $wR$  factor = 0.062  
Data-to-parameter ratio = 24.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(2R)-N-[3-[Bis(methylsulfanyl)methyleneamino]-  
propionoyl]bornane-10,2-sultam**In the title compound,  $\text{C}_{16}\text{H}_{26}\text{N}_2\text{O}_3\text{S}_3$ , adjacent molecules are  
arranged in a head-to-tail manner and linked *via* weak  $\text{C}-\text{H}\cdots\text{S}$   
interactions, forming helical chains parallel to the  $a$  axis.  
These chains are connected *via*  $\text{C}-\text{H}\cdots\text{O}$  interactions.**Comment** $\beta$ -Amino acids are a key part of many natural products  
displaying antibiotic, antifungal and cytotoxic properties  
(Cardillo & Tomasini, 1996), precursors of  $\beta$ -lactam antibiotics  
(Georg, 1993), and peptidomimetics showing remarkable  
resistance for enzymatic degradation (Steer *et al.*, 2000).  
 $\beta$ -Amino acids are also very interesting because of their  
tendency to form highly stable helical structures: 14-helix, 12-  
helix, 10/12-helix for penta-, hexa- and heptamers, composed  
of appropriately substituted  $\beta$ -amino acids (Seebach *et al.*,  
1998; Cheng *et al.*, 2001). The stability of these structures is  
higher than those of similar ones formed by  $\alpha$ -amino acids.Even though  $\beta$ -amino acids can be obtained from natural  
sources and *via* chemical synthesis (Liu & Sibi, 2002), taking  
into account their significance and their further applications as  
peptide mimetics, there is a need for elaboration of new,  
effective and stereoselective methods for their preparation.  
One of the possible synthetic routes leading to the above  
compounds involves synthesis of a simple precursor (*e.g.*  $\beta$ -  
alanine) attached to a chiral auxiliary group. Among many  
chiral auxiliaries, bornane-10,2-sultam is one of the most  
popular and effective groups to create new chiral centres. This  
group was introduced by Oppolzer *et al.* (1984), and hence-  
forth has been widely used for the synthesis of many  
compounds (Oppolzer, 1990; Szymańska *et al.*, 2000). We  
applied the Schiff base obtained from  $\beta$ -alanine and Oppol-  
zer's sultam for the synthesis of some  $\beta^2$ -substituted  $\beta$ -amino  
acids (Woźniak *et al.*, 2006). For this purpose we modified the  
methodology described by Ponsinet *et al.* (2000). The  
precursor was chosen on the basis of the simplicity of its  
synthesis, the very high asymmetric induction characteristic  
for the sultam group and the high tendency of  $N$ -acylsultam  
derivatives to be crystalline.In the title compound, (I), bond lengths and angles are  
typical for substituted bornane-10,2-sultam (Hughes *et al.*,Received 4 November 2005  
Accepted 12 December 2005  
Online 21 December 2005We dedicate this paper to the  
memory of Professor Leszek  
B. Łankiewicz (1960–2003),  
an outstanding scientist and  
magnificent person, who was  
our inspiration in so many  
ways.



**Figure 1**  
The molecular structure of (I), showing the atom-labelling scheme and 50% probability displacement ellipsoids.



**Figure 2**  
The packing of (I). The C—H...O and C—H...S interactions are represented by dashed lines. H atoms not involved in these interactions have been omitted.

1999; Raczko *et al.*, 2000; Szymański *et al.*, 2001) (Fig. 1 and Table 1). The cyclohexane ring adopts a fairly regular boat conformation, with ring puckering parameters  $Q = 0.993(2) \text{ \AA}$ ,  $\theta = 88.65(7)^\circ$  and  $\varphi = 355.21(7)^\circ$  (Cremer & Pople, 1975) and asymmetry parameter  $\Delta C_s(C1) = 9.4(1)^\circ$  (Duax *et al.*, 1976). Adjacent molecules of (I) are arranged in a head-to-tail manner and linked *via* weak C—H...S interactions (Table 2), forming helical chains parallel to the *a* axis (Fig. 2). These chains are connected *via* C—H...O interactions (Table 2).

## Experimental

Compound (I) was synthesized from bornane-10,2-sultam and methyl 3-[[bis(methylthio)methylene]amino]propionate, according to the

procedure given by Oppolzer *et al.* (1989). Such a compound can, after enolization by treatment with either *n*-butyllithium or lithium diisopropylamide at 195 K, be alkylated by selected alkyl halides yielding precursors of  $\beta^2$ -amino acids (superscript 2 denotes the position of the introduced group  $\alpha$  with reference to the carboxyl group). Crystals of (I) suitable for X-ray study were grown from an ethanol solution by slow evaporation at room temperature (m.p. 363–364 K).

## Crystal data

$C_{16}H_{26}N_2O_3S_3$   
 $M_r = 390.57$   
Orthorhombic,  $P2_12_12_1$   
 $a = 7.742(2) \text{ \AA}$   
 $b = 12.348(3) \text{ \AA}$   
 $c = 19.405(3) \text{ \AA}$   
 $V = 1855.1(7) \text{ \AA}^3$   
 $Z = 4$   
 $D_x = 1.398 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation  
Cell parameters from 16987 reflections  
 $\theta = 3\text{--}30^\circ$   
 $\mu = 0.42 \text{ mm}^{-1}$   
 $T = 100(2) \text{ K}$   
Prism, colourless  
 $0.7 \times 0.5 \times 0.3 \text{ mm}$

## Data collection

Kuma KM-4 CCD diffractometer  
 $\omega$  scans  
Absorption correction: none  
23438 measured reflections  
5320 independent reflections  
5197 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$   
 $\theta_{\text{max}} = 30.0^\circ$   
 $h = -10 \rightarrow 10$   
 $k = -17 \rightarrow 17$   
 $l = -27 \rightarrow 25$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.025$   
 $wR(F^2) = 0.062$   
 $S = 1.08$   
5320 reflections  
221 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0336P)^2 + 0.344P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.007$   
 $\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$   
Absolute structure: Flack (1983),  
2253 Friedel pairs  
Flack parameter: 0.02 (4)

**Table 1**  
Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

N7—C8	1.3921 (14)	C12—S14	1.7780 (13)
N7—S21	1.7019 (10)	C20—S21	1.7880 (12)
C10—N11	1.4737 (16)	S21—O22	1.4353 (11)
N11—C12	1.2700 (16)	S21—O23	1.4395 (10)
C12—S13	1.7738 (13)		
C1—C2—C3—C4	−3.28 (12)	C1—C6—N7—S21	26.16 (11)
C20—C1—C6—N7	−33.23 (12)	C5—C6—N7—S21	140.87 (8)
C4—C5—C6—C1	−7.82 (11)	S21—N7—C8—C9	−24.08 (15)

**Table 2**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
C3—H3A...S14 <sup>i</sup>	0.99	2.95	3.484 (2)	115
C20—H20B...O24 <sup>ii</sup>	0.99	2.28	3.242 (2)	164

Symmetry codes: (i)  $x - \frac{1}{2}, -y + \frac{3}{2}, -z + 2$ ; (ii)  $x - 1, y, z$ .

All H atoms were positioned geometrically and refined as riding atoms, with C—H distances of 0.98–1.00  $\text{\AA}$ , and with  $U_{\text{iso}}(\text{H})$  values of  $1.5U_{\text{eq}}(\text{C})$  (for methyl H atoms) or  $1.2U_{\text{eq}}(\text{C})$  (for the remaining H atoms). The absolute configuration based on the Flack (1983) parameter is consistent with that of bornane-10,2-sultam (Oppolzer, 1990; Kiegiel & Jurczak, 1999).

Data collection: *KM-4 CCD Software* (Oxford Diffraction, 2003); cell refinement: *KM-4 CCD Software*; data reduction: *KM-4 CCD Software*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

This work was supported by the Gdańsk University grant DS/8350-5-0131-5.

## References

- Cardillo, G. & Tomasini, C. (1996). *Chem. Soc. Rev.* pp. 117–128.
- Cheng, R. P., Gellman, S. H. & DeGrado, W. F. (2001). *Chem. Rev.* **101**, 3219–3232.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Duax, W. L., Weeks, C. M. & Röhrer, D. C. (1976). *Topics in Stereochemistry*, Vol. 9, edited by E. L. Eliel & N. Allinger, pp. 271–383. New York: John Wiley.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Georg, G. I. (1993). *The Organic Chemistry of  $\beta$ -Lactams*, edited by G. I. Georg. New York: Verlag Chemie.
- Hughes, A. B., Mackay, M. F. & Mccaffrey, N. L. (1999). *Acta Cryst.* **C55**, 610–612.
- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Kiegiel, K. & Jurczak, J. (1999). *Tetrahedron Lett.* **40**, 1009–1012.
- Liu, M. & Sibi, M. P. (2002). *Tetrahedron*, **40**, 7991–8035.
- Oppolzer, W. (1990). *Pure Appl. Chem.* **62**, 1241–1250.
- Oppolzer, W., Chapuis, C. & Bernardinelli, G. (1984). *Helv. Chim. Acta*, **67**, 1397–1398.
- Oppolzer, W., Moretti, R. & Thomi, S. (1989). *Tetrahedron Lett.* **30**, 6009–6010.
- Oxford Diffraction (2003). *KM-4 CCD Software*. Version 1.171. Oxford Diffraction Poland, Wrocław, Poland.
- Ponsinet, R., Chassaing, G., Vaissermann, J. & Lavielle, S. (2000). *Eur. J. Org. Chem.* 83–90.
- Raczko, J., Achmatowicz, M., Kwiatkowski, P., Chapuis, C., Urbańczyk-Lipkowska, Z. & Jurczak, J. (2000). *Tetrahedron Asymmetry*, **11**, 1027–1041.
- Seebach, D., Abele, S., Gademan, K., Guichard, G., Hinterman, T., Jaun, B., Matthews, J. L. & Schreiber, J. V. (1998). *Helv. Chim. Acta*, **81**, 932–982.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Steer, D. L., Lwe, R. A., Perlmutter, P., Smith, A. I. & Auilar, M. I. (2000). *J. Peptide Sci.* **6**, 470–477.
- Szymańska, A., Wiczerzak, E. & Łankiewicz, L. (2000). *Wiad. Chem.* **54**, 759–791. (In Polish.)
- Szymański, S., Chapuis, C. & Jurczak, J. (2001). *Tetrahedron Asymmetry*, **12**, 1939–1945.
- Woźniak, D., Szymańska, A., Ołdziej, S., Łankiewicz, L. & Grzonka, Z. (2006). *Pol. J. Chem.* In the press.